

PROCESS FOR ADVANCED TREATMENT FOR
DESULFURIZATION OF EXHAUST GAS WITH SEAWATER

5 BACKGROUND OF THE INVENTION

1. Field of the Invention

The present invention relates to a method of wet removal of an acidic component in an exhaust gas with seawater, in particular an acidic component exhausted from a boiler or various furnaces, and an acidic component from an exhaust gas containing sulfite gas.

10 2. Description of the Related Art

At present, for example, when heavy oil, coal, etc. are burned in combustion equipments such as a boiler or various types of furnaces and since the fuels contain a sulfur component, the sulfur component is combined with the oxygen in the air at the time of combustion to form sulfur oxides which end up included in the combustion exhaust gas. Such sulfur oxides travel with the gas from the source of generation to regions as much as several thousand kilometers away to cause phenomena such as acid rain or acid fog, polluting the air, water, soil, etc. over a vast region and having detrimental effects on human health.

25 However, along with the increased international interest in global environmental issues, global scale measures have been sought. In Japan, along with advances made in exhaust gas desulfurization or denitrogenation technologies etc., efficient boiler combustion technology has been accomplished. Substantially 100% antipolution measures are now taken at the source of generation. For example, as a method of wet removal of the acidic component such as the sulfur oxides contained in the gas, a packed column, spray column, bubble cap column, perforated or grid plate column without weir and downcomer etc. have been used to bring the gas to be treated and an alkali treatment solution countercurrently

into contact. That is, it can be said that the desulfurization treatment technology for environmental protection has reached the almost perfected stage. For example, a desulfurization efficiency of 90 to 99% has 5 been obtained and commercialized. However, since calcium hydroxide, calcium carbonate, sodium hydroxide, magnesium hydroxide, etc. are used as the alkali treatment solution, there occur problem that, in addition to higher cost, the waste solutions have to be treated, the solid 10 substances have to be treated, etc. Thus, there are problems of complication of the process and high construction and running costs. Therefore, as a method of treatment of exhaust gas using seawater, the present inventors developed a method of wet treatment of exhaust 15 gas by using a perforated or grid plate column without weir and downcomer in JP-A-11-290643 (method of treatment of acidic component in an exhaust gas with seawater). Nevertheless, this method was insufficient in the low Ug 20 region.

As explained above, even a process for removing an acidic component from an exhaust gas, developments of new technologies for simplification of the process, making the equipment more compact, and sharply reducing the construction and running costs are sought not only in 25 the industrialized countries, but in the developing countries as well. In particular, a reliable process is sought for cases where advanced treatment of all regions, including the low Ug region, is frequently required.

SUMMARY OF THE INVENTION

Accordingly, objects of the present invention are to 30 develop new technologies capable of highly removing an acidic component such as a sulfate gas contained in an exhaust gas with a simplified process in a more compact equipment, and at a remarkably reduced the construction and running costs, in view of the international needs, at 35 a wide region including a low Ug and low L/G region.

In accordance with the present invention, there is

provided a method for removing an acidic component contained in an exhaust gas comprising by using a system comprising (a) a gas-liquid contact apparatus composed of an absorption column provided internally with at least 5 one perforated plate at the top, bottom, or both top and bottom of the absorption column packed with at least one type of fillers, (b) an apparatus for introducing seawater to the absorption column, (c) an apparatus for oxidizing the seawater after gas-liquid contact, and (d) 10 an apparatus for mixing a noncontact seawater with the seawater after subjected to the mixing and oxidation, whereby the exhaust gas containing an acidic component is brought into gas-liquid contact with the seawater.

In accordance with the present invention, there is 15 also provided a wet method for removing an acidic component contained in an exhaust gas, in a gas-liquid contact apparatus including an absorption column having a column diameter of at least 500 mm and provided with at least one perforated plate having an free-space ratio F_c 20 of 0.25 to 0.5 and provided with at least one type of packing material with a packing height of 0.5 m to 4 m, comprising

supplying seawater in such an amount that a ratio L/G of the flow rate L ($\text{kg}/\text{m}^2 \cdot \text{hr}$) of the seawater 25 to the flow rate G ($\text{kg}/\text{m}^2 \cdot \text{hr}$) of the gas to be treated from the top of the column is at least 3.6 and a flow rate L of the seawater is 1×10^4 to 25×10^4 $\text{kg}/\text{m}^2 \cdot \text{hr}$ and introducing a treated gas in such an amount that a range 30 of a superficial gas velocity U_g in the apparatus from the bottom of the gas-liquid contact apparatus is 0 to $2 \cdot U_{gm}$ (m/sec):

in the case of using a perforated or grid plate column without weir and downcomer composed of at least one perforated plate and the ratio ρ_g/ρ_L of the density ρ_g 35 (kg/m^3) of the treated gas to the density ρ_L (kg/m^3) of seawater of 1030 is at least 0.838×10^{-3} ,

CONTINUATION-RECEIVED

$$U_{gm} = 49.14 Fc^{0.7} (\rho_g / \rho_L \times 10^{-3})^{-0.5} \cdot (L/G)^{-1/3} \cdot \sqrt{g \cdot L}$$

wherein L is a capillary constant $\sqrt{2\sigma/\rho_L \cdot g}$,

g is a gravitational acceleration (m/sec^2), and

σ is a surface tension of seawater (kg/sec^2)

- 5 whereby the gas to be treated and seawater are countercurrently brought into gas-liquid contact.

BRIEF DESCRIPTION OF THE DRAWINGS

The present invention will be better understood from the description set forth below with reference to the 10 accompanying drawings, wherein:

Fig. 1 is a schematic view showing the relationship between the flow rate L of the treatment liquid and the superficial gas velocity U_g in the operating region according to the method of the present invention, wherein 15 the regions A and B show the operating regions according to Japanese Examined Patent Publication (Kokoku) No. 51-31036 and Japanese Examined Patent Publication (Kokoku) No. 60-18206, while the region C shows the operating region according to JP-A-11-290643. The operating region in the present invention is the region of U_g of more than 20 0, but not more than $2 \times U_{gm}$, surrounded by the lines of the flow rate of more than 1×10^4 ($kg/m^2 \cdot hr$), $L/G=3.6$, and the flow rate 25×10^4 ($kg/m^2 \cdot hr$), of the treatment liquid of Fig. 1.

25 Fig. 2 is a view of the treatment system centered around the apparatus for removal of the acidic component in the exhaust gas with seawater, that is, a gas-liquid contact apparatus, according to the present invention.

30 ~~Fig. 3 is an example of an absorption apparatus consisting of a combination of a perforated or grid plate without weir and downcomer and a packed column in the case of treating boiler exhaust gas by the method according to the present invention.~~

35 ~~Fig. 4 is another example of an absorption apparatus consisting of a combination of a perforated or grid plate without weir and downcomer and a packed column in the~~

(Ans a27)

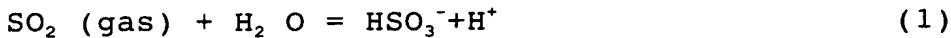
~~An~~ case of treating boiler exhaust gas by the method according to the present invention.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

According to the present invention, as shown in Fig. 2, since an absorption column composed of a combination of a perforated or grid plate without weir and downcomer and packing material is used, operation is possible from a small liquid/gas ratio to a large liquid/gas ratio, the alkali contained in the used cooling seawater or alkaline waste seawater used in power plants, factories, etc. can be utilized, and the desired acidic component can be removed, without using any chemicals at all, for the purpose of absorbing the gas and of adjusting pH of the absorbing liquid, in the entire process. Note that here the perforated or grid plate without weir and downcomer means a perforated or grid plate having a free-space ratio F_c of 0.25 to 0.5, preferably 0.3 to 0.4. In the present invention, by using a gas-liquid contact apparatus comprising a combination of at least one such perforated or grid plate and packing material with a packing height of at least 0.5 m, preferably 0.5 to 4 m, it is possible to use seawater or waste seawater to remove, in a wet manner, acidic components such as sulfite gas contained in an exhaust gas. Further, since the seawater after the gas-liquid contact makes the components absorbed into it harmless by the supply of air and oxidation by a seawater treatment system, there is no problem with pollution of the ocean at all.

The present invention is not limited in content to the following mechanism, but the chemical principle of the seawater desulfurization process according to the present invention is now explained, when the acidic component is sulfite gas (SO_2).

That is, the sulfite gas (SO_2) contained in the exhaust gas is absorbed in the seawater, and then converted to bisulfite ions as shown in formula (1):



The bisulfite ion thus formed is converted to a sulfate ion as shown in formula (2) by aeration oxidation:



5 The hydrogen ion thus produced by the reaction formula (2) is then reacted with a carbonate ion or bicarbonate ion contained in seawater and is thus neutralized as shown in the reaction formulae (3) and (4):



10 According to the present invention, the sulfite gas finally becomes a sulfate ion which is then dissolved into the seawater. The sulfur content in the ocean is said to be as high as 10^{15} tons. This corresponds to about 2300 mg/L of a sulfate ion. The concentration of the harmless sulfate ion contained in the finally treated seawater effluent from the present process only increases by about several mg/liter, and therefore the increase in 15 the sulfur content in the ocean is extremely slight.

20 Some of the sulfur derived from the sulfur oxides produced due to the use of fossil fuels in modern day industry is naturally recycled by passing through the atmosphere and returning to the earth or the sea in the form of acid rain. As opposed to this, the method of removal of the acidic component in a gas in the present invention is an environmentally friendly technique. According to the process of the present invention, it is possible to recycle sulfur to the ocean in this way by a 25 short cut and effectively prevent the damages such as acid rain, air pollution.

30 *(Ans a3)* ~~As explained above, the present invention successfully uses a gas-liquid contact apparatus equipped inside thereof with an absorption column containing at least one perforated plate and at least one type of packing materials with a packing height of at least 0.5 m, preferably 0.5 m to 4 m (e.g., Raschig rings, pole~~

(Ans 93)

5 rings, Terralets, interlock saddles, etc.) and introduces
10 seawater from the top of the gas-liquid contact apparatus
15 or introduces seawater from the top of the gas-liquid
20 contact apparatus so as to cause countercurrent gas-
25 liquid contact with the gas to be treated and to use the
30 alkali in the seawater to effectively and advancely
35 remove the sulfur oxides contained in the exhaust gas.

(Ans 94)

10 Note that the technology for treatment of an exhaust
15 gas using a perforated plate or grid plate column without
20 weir and downcomer (i.e. "Moretana" column) is disclosed
25 in Japanese Examined Patent Publication (Kokoku) No. 51-
30 31036 and Japanese Examined Patent Publication (Kokoku)
35 No. 60-18208 (or U.S. Patent Nos. 3892837 and 3941572)
but the present inventors found that the operating
regions A and B (see Fig. 1) already shown there is not
suited for the treatment of exhaust gas utilizing
seawater. The above-mentioned JP-A-11-290643 proposed the
need for the ratio L/G of the flow rate of gas G supplied
to the column and the flow rate L of seawater to be at
least 3.6, preferably 7 to 25, and for the superficial
gas velocity U_g passing through the "Moretana" column and
the flow rate L of the treatment liquid to be the
relation in the region C of Fig. 1, that is, in the range
from more than $3.43 L^{-0.0807}$ Ugm (m/sec) to 8 (m/sec).
However, the outside of this range, particularly when U_g
is too low, there is the problem that the efficiency of
gas-liquid contact drastically falls and it becomes
impossible to remove the acidic component. However,
according to the present invention, by utilizing an
absorption apparatus composed of a combination of a
"Moretana" column and a packing column, the desired gas-
liquid absorption is effectively performed even in the
region D of Fig. 1 and the highly efficient
desulfurization treatment becomes possible.

35 Seawater contains approximately 110 to 130 mg/L, in
terms of CaCO_3 , of alkali. The present invention makes
highly effective use of seawater - which can be easily

DRAFTED BY ETSU

obtained in large quantities in the area near the sea. According to the present invention, for example, in the case of a power station using seawater of the nearby ocean as cooling water, it is possible to reuse the
5 seawater after the cooling, originally to be returned to the ocean, to treat the exhaust gas from a boiler and remove the sulfur oxides contained in the exhaust gas therefrom at a high desulfurization rate. Furthermore, according to the present invention even in the case of a
10 factory producing magnesium hydroxide from seawater or a pulp and paper factory using seawater, it is possible to reutilize the waste seawater, which originally has to be retreated before discharge to the ocean, and to use the
15 above gas-liquid apparatus to treat both the acidic exhaust gas and waste seawater.

According to the present invention, the sulfur-absorbed liquid is discharged into the ocean after the acidic sulfur-absorbed liquid containing bisulfite ion, which becomes a source of COD (i.e., chemical oxygen demand), is oxidized by air in the aeration oxidation vessel and after the pH of the mixing seawater is recovered by decarbonization, and therefore, it is possible to discharge the seawater into the ocean and recover the quality of the seawater without using any
25 chemicals.

EXAMPLES

The present invention will now be further explained in detail by, but is by no means limited to, the following Examples.

Example 1

Figure 3 shows an example of a flow of the process, where boiler exhaust gas is treated by the method of the present invention.

Combustion exhaust gas containing about 1000 ppm of
35 sulfur oxides exhausted from the boiler 1 is introduced into an electric dust collector 2 to remove the dust, then the exhaust gas is passed through a gas-liquid

contact apparatus 3 provided with one perforated plate having a free-space ratio F_c of 0.3 and a packed portion having 2 m of a packing material (i.e., a ring type, dimensions: 100 x 78 x 32 mmH) for treatment, then
5 dispersed therefrom, as a gas effluent, into the atmosphere. Inside the gas-liquid contact apparatus 3, the exhaust gas is introduced from the bottom, while seawater pumped out from the sea is introduced from the top of the gas-liquid contact apparatus 3, whereby the seawater is countercurrently into contact with the
10 exhaust gas in the apparatus 3 and absorb and remove the sulfur oxides contained in the exhaust gas. The acidic waste seawater containing the bisulfite ion discharged from the bottom of the apparatus, as a liquid effluent,
15 is oxidized with air in the air oxidation vessel 4 to oxidize the sulfite ion and to decarbonize the gas effluent, whereby the pH of seawater is recovered and, then the seawater is discharged to the ocean. The removal rate was 90 to 99% with a liquid gas ratio L/G of 5 to
20 10, a Ug of 1 to 2 m/sec, and a concentration of sulfur oxides in the gas to be treated of 10 to 100 ppm.

Example 2

Figure 4 is another example of the method of treating boiler exhaust gas by the method of the present invention and shows a flowchart of an example in the case of treatment while maintaining a high desulfurization rate corresponding to the low duty operation of a boiler.
25

Combustion exhaust gas containing about 800 ppm of sulfur oxides exhausted from the boiler 1 is introduced
30 into an electric dust collector 2 to remove the dust, then the exhaust gas is passed through a gas-liquid contact apparatus 3 provided with two perforated plates having a free-space ratio F_c of 0.3 and a packed portion having 1.5 m of a packing material (i.e., ring type, dimension: 100 x 78 x 32 mmH) for treatment, then
35 discharged into the atmosphere. Inside the gas-liquid contact apparatus 3, the exhaust gas is introduced from

the bottom thereof, while the seawater pumped out from
the sea is introduced from the top thereof, whereby the
seawater is brought into countercurrent contact with the
exhaust gas in the apparatus 3 and the sulfur oxides in
5 the exhaust gas are absorbed and removed. The acidic
waste seawater containing the bisulfite ion discharged
from the bottom of the apparatus as a liquid effluent is
oxidized and decarbonized by air in the air oxidation
vessel 4, and the pH of seawater is recovered, then the
10 seawater is discharged, as a liquid effluent, to the
ocean. The removal efficiencies were 99.4%, 98.7%, 97.9%,
96.9%, and 92.5% with a flow rate of seawater L (5×10^4
 $\text{kg/m}^2\cdot\text{hr}$)₂Ug of 0.25, 0.5, 1, 1.5, and 2 m/sec and a
concentration of sulfur oxides in the gas to be treated
15 of 5, 10, 17, 25, and 60 ppm.

(Ans A5)

As explained above, according to the present
invention, it is possible to advanceley and effectively
treat the sulfur oxides contained in an exhaust gas with
seawater, despite the region being unable to be treated
20 by "Moretana" column in the past, that is, a low Ug, and
possible to protect the environment simply and with a
compact equipment and low cost.